# Slide 1

Chemistry: A Molecular Approach, 1st Ed. Nivaldo Tro **Chapter 13 Chemical Kinetics** Roy Kennedy Massachusetts Bay Community College Wellesley Hills, MA 2008, Prentice Hall

#### Slide 2

- Catalysts

  catalysts are substances that affect the rate of a reaction without being consumed

  catalysts work by providing an alternative mechanism for the reaction

  with a lower activation energy

  catalysts are consumed in an early mechanism step, then made in a later step

mechanism without catalyst

mechanism with catalyst

 $\mathrm{O}_{3(g)} + \mathrm{O}_{(g)} \rightarrow \ 2 \ \mathrm{O}_{2(g)} \quad \text{V. Slow} \quad \mathrm{Cl}_{(g)} + \mathrm{O}_{3(g)} \Leftrightarrow \mathrm{O}_{2(g)} + \mathrm{ClO}_{(g)} \quad \text{Fast}$ 

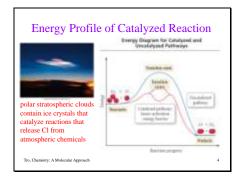
 $\text{ClO}_{(g)} + \text{O}_{(g)} \rightarrow \text{ O}_{2(g)} + \text{Cl}_{(g)} \quad \text{Slow}$ 

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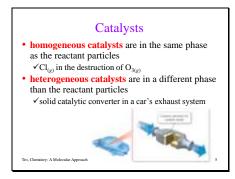
#### Slide 3

# Ozone Depletion over the Antarctic Tro, Chemistry: A Molecular Approach

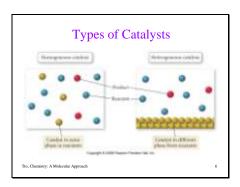
# Slide 4



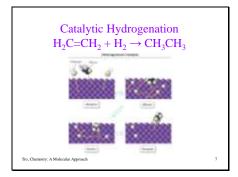
# Slide 5



### Slide 6



# Slide 7



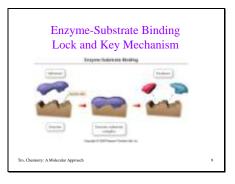
#### Slide 8

#### Enzymes

- because many of the molecules are large and complex, most biological reactions require a catalyst to proceed at a reasonable rate
- protein molecules that catalyze biological reactions are called **enzymes**
- enzymes work by adsorbing the substrate reactant onto an active site that orients it for reaction

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#### Slide 9




Slide 10

